POLYMER SUPPORTED DNA SYNTHESIS USING HYDROXYBENZOTRIAZOLE ACTIVATED PHOSPHOTRIESTER INTERMEDIATES

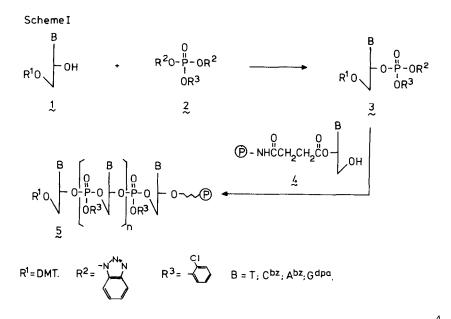
J.E. Marugg[#], L.W. McLaughlin^{+*}, N. Piel⁺, M. Tromp[#], G.A. van der Marel[#] and J.H. van Boom^{#*} [#]Gorlaeus Laboratories, State University of Leiden, P.O. Box 9502, 2300 RA Leiden, The Netherlands ⁺Max-Planck-Institut für experimentelle Medizin, Abteilung Chemie, Hermann-Rein-Str. 3, D-3400 Göttingen, GFR

ABSTRACT

In the present method appropriate N-protected-5'-O-(4,4'-dimethoxytrityl)-2'-deoxynucleosides are phosphorylated with the reagent 2-chlorophenyl-O,O-bis(1-benzotriazolyl)-phosphate and the 3'-phosphorylated intermediates thus obtained are used directly to elongate the polymer support Kieselgur-polydimethylacrylamide. According to this procedure, we prepared two DNA fragments: one decamer and a heptadecamer.

One of the essential steps in the synthesis of DNA on a solid support is the introduction of an intermediate 3'-5'-internucleotide phosphotriester linkage. In one of the existing approaches, i.e., the phosphotriester approach¹, the formation of a triester linkage between an immobilized DNA fragment and an incoming nucleotide is achieved as follows: The immobilized DNA fragment with a free 5'-OH (e.g., <u>4</u> in Scheme 1) is coupled, in the presence of an activating agent, with a suitably protected 2'-deoxynucleoside carrying a 3'-phosphodiester function (e.g., <u>3</u>, $\mathbb{R}^2 = H$). An alternative route to achieve the same goal would be to circumvent not only the use of the 3'-phosphodiester intermediate <u>3</u> ($\mathbb{R}^2 = H$), but also the activating agent.

We now wish to report that the bifunctional phosphorylating agent 2-chlorophenyl-0,0-bis(1--benzotriazolyl)phosphate² (2) can be used as an alternative for the introduction of triesterlinkages between an immobilized DNA fragment and an incoming suitably protected 2'-deoxynucleoside. The route we have employed is illustrated in Scheme 1. The intermediate phosphotriester 3, which will be coupled directly with the immobilized deoxynucleoside 4, was obtained by the following procedure. The 5'-protected 2'-deoxynucleoside 1 (R¹ = 4,4'-dimethoxytrityl; 1.0 mmol) was dissolved in anhydrous pyridine and evaporated to give an oil. An equimolar amount of phosphorylating agent 2 withdrawn from a stock solution of 2 (0.2 M) in dioxane³ was added to the above oil. TLC-analysis (CHCl₃-MeOH; 92:8, v/v), after 15 min at 20°C,



showed complete conversion of starting material <u>1</u> into a product with zero mobility⁴. A stock solution of <u>3</u> (0.2 M) prepared in dioxane can be stored for several weeks at -20°C without any appreciable loss of activity. We have examined a number of the presently available polymer supports for <u>DNA</u> synthesis by the present approach⁵. It would appear that the polymer Kiesel-gur-polydimethylacrylamide^{1a,b,6}, in which the first deoxynucleoside is immobilized by a succinate linkage to the solid support (capacity: 83 μ mol/g), is most effective.

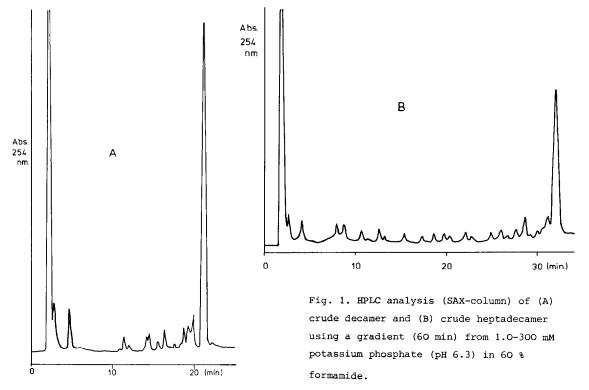
In a typical elongation process, intermediate <u>3</u> (0.1 mmol) was mixed with N-methylimidazole/ N,N-diisopropylethylamine/dioxane⁷ (2:1:1, v/v, 100 μ l) and injected using a glass syringe, onto the solid support <u>4</u> (120 mg) packed into a column which is part of a continuous flow--bench synthesizer^{1b}, and left in contact with the support for 15 min at 20°C. The effective-

Table	1.	Deprotection,	coupling	and	wash	cycle.	

Steps	Solvents and reagents Time (m	nin)
1.	10 % Trichloroacetic acid/1,2-dichloroethane	3
2.	DMF	2
3.	Pyridine	2
4.	Dioxane	5
5.	Coupling (stop flow)	15
6.	Dioxane	2
7.	Capping (Ac ₂ O/Et ₃ N/N-methylimidazole/dioxane: 4.5:4.5:1:30, v/v)	5
8.	Dioxane	3
9.	1,2-Dichloroethane	3

ness of the coupling process was monitored by measuring the release of the DMT-cation by UV spectroscopy at 498 nm. Based on the spectroscopic analysis the coupling process proceeds with an efficiency of greater than 95 % (based on <u>4</u>). A complete elongation step, which includes the removal of the DMT group, the introduction of a 3'-5'-phosphotriester bond and subsequent capping of unreacted 5'hydroxyl functions is illustrated in Table 1. From the steps described in Table 1 one complete cycle (i.e., conversion of <u>5</u>: n = 0; $R^1 = DMT$) into 5: n = 1; $R^1 = DMT$) can be performed in 40 min.

In applying the present elongation procedure, we prepared two immobilized DNA fragments: a decamer (i.e., 5; n = 8; sequence d(ApTpTpApTpTpApTpTpG)) and an heptadecamer (i.e., 5; n = 15; sequence d(ApApGpGpTpGpCpCpCpTpGpTpTpCpApTpC)). The fully protected and immobilized DNA fragments were deblocked and released from the solid support by a three step procedure. In the first step, the aryl protective groups (\mathbb{R}^3) of the phosphotriesters and the succinate linkages, were removed by treating the immobilized fragments with (E)-2-nitrobenzaldehyde oxime⁸ and N¹,N¹,N³,N³-tetramethylguanidine in dioxane/water. The N-acyl protective groups [i.e., benzoyl (bz) and diphenylacetyl (dpa)] were deblocked by a short treatment with aqueous ammonia for 24 h at 50°C. Finally, the dimethoxytrityl groups were deblocked by a short treatment with aqueous acetic acid. The HPLC analyses^{1b,9} of the crude DNA fragments obtained for the decamer and heptadecamer are illustrated in Figures 1A and 1B respectively. It can be seen that the hydroxybenzotriazole (HOBT) approach affords DNA fragments of rather high quality. The two purified¹⁰ (Sephadex G50) DNA fragments were completely digested, as judged from HPLC analysis, by venom and spleen phosphodiesterase to give the expected 2'-deoxynucleo-tides in the correct ratios.



The results obtained above indicate that the HOBT approach is in several aspects superior over the generally accepted phosph otriester procedure. For instance, the preparation of the key intermediate 3 is straightforward and does not involve a time consuming purification step. Furthermore, no activating agents [e.g., mesitylenesulfonyl-3-nitro-1,2,4-triazole¹¹ (MSNT)], which may give rise to side reactions with thymidine 12 or guanosine 13, are required. Another interesting feature of the HOBT approach is that the coupling step (i.e., conversion of 5; n = 0 to 5; n = 1) is nearly three times faster than the same step in a phosphotriester approach in which 3 (R^2 = triethylammonium salt) is coupled in the presence of the well established activating agent MSNT^{1a}. The HOBT coupling time is comparable with that required for the introduction of a phosphotriester linkage by the phosphite-triester $approach^{14}$.

In conclusion, we believe¹⁵ that the HOBT approach will be an economical and simple procedure for the synthesis of DNA fragments on a solid support.

References and notes

- 1. a) M.J. Gait, H.W.D. Matthes, M. Sing, B.S. Sproat and R.C. Titmas, Nucleic Acids Res. 10, 6243 (1982). b) M.J. Gait, H.W.D. Matthes, M. Sing, B.S. Sproat and R.C. Titmas in "Chemical and Enzymatic Synthesis of Gene Fragments", ed. H.G. Gassen and A. Lang, Verlag Chemie, Weinheim (1982) pp. 1-42. c) H. Ito, Y. Ike, S. Ikuta and K. Itakura, Nucleic Acids Res. 10, 1755 (1982). d) T. Horn, M. Vasser, M.E. Struble and R. Crea, Nucleic Acids Res. s7, 225 (1980). e) H. Köster, A. Stumpe and A. Wolter, Tetrahedron Lett. 24, 747 (1983). f) G.A. van der Marel, J.E. Marugg, E. de Vroom, G. Wille, M. Tromp, C.A.A. van Boeckel and J.H. van Boom, Recl. Trav. Chim. Pays-Bas 101, 234 (1982).
- 2. a) G.A. van der Marel, C.A.A. van Boeckel, G. Wille and J.H. van Boom, Tetrahedron Lett. 22, 3887 (1981). b) J.H. van Boom, G.A. van der Marel, C.A.A. van Boeckel, G. Wille and C.F. Hoyng in "Chemical and Enzymatic Synthesis of Gene Fragments", ed. H.G. Gassen and A. Lang, Verlag Chemie, Weinheim (1982) pp. 53-70.
- 3. The phosphorylating agent 2 can be kept, under the exclusion of moisture, for several months at $0^{\circ}C$ without any appreciable loss of activity. This observation was also supported by ¹P-NMR studies: J.E. Marugg et al., to be
- 4. This observation was also supported by published.
- 5. J.E. Marugg et al., to be published. See also J.E. Marugg, G.A. van der Marel, E. de Vroom, D. Bosch and J.H. van Boom, Recl. Trav. Chim. Pays-Bas 101, 411 (1982).
- 6. R.C. Sheppard, Chemistry in Britain, 402 (1983).
- The presence of this tertiary base has a dramatic effect on the rate of coupling. The 7. reason for this is presently under investigation.
- 8. C.B. Reese and L. Zard, Nucleic Acids Res. 9, 4611 (1981).
- 9. L.W. McLaughlin and J.U. Krusche in "Chemical and Enzymatic Synthesis of Gene Fragments", ed. H.G. Gassen and A. Lang, Verlag Chemie, Weinheim (1982) pp. 177-198.
- 10. J.F.M. de Rooij, G. Wille-Hazeleger, P.H. van Deursen, J. Serdijn and J.H. van Boom, Recl. Trav. Chim. Pays-Bas, <u>98</u>, 537 (1979).
- 11. S.S. Jones, B. Rayner, C.B. Reese, A. Ubasawa and M. Ubasawa, Tetrahedron 36, 3075 (1980).
- 12. E. Uhlmann and W. Pfleiderer, Helv. Chim. Acta 64, 1688 (1981).
- 13. C.B. Reese and A. Ubasawa, Tetrahedron Lett. 21, 2265 (1980).
- 14. a) M.D. Matteucci and M.H. Caruthers, Tetrahedron Lett. 21, 719 (1980). b) M.D. Matteucci and M.H. Caruthers, J. Am. Chem. Soc. 103, 3185 (1981). c) M.H. Caruthers in "Chemical and Enzymatic Synthesis of Gene Fragments", ed. H.G. Gassen and A. Lang, Verlag Chemie, Weinheim (1982) pp. 71-79.
- 15. Preliminary experiments showed that the HOBT approach could also be applied on a Synthesis Automation Machine.

(Received in Germany 20 June 1983)

3992